Application No.: 10/511,715

Art Unit: 1793

Amendment under 37 CFR §1.111

Attorney Docket No.: 042834

AMENDMENTS TO THE CLAIMS

The listing of claims below replaces all prior versions of claims in the application.

1. (Previously Presented): A $12\text{CaO} \cdot 7\text{Al}_2\text{O}_3$ compound, which incorporates a negative hydrogen ion (H⁻, H²⁻, H₂⁻) at a concentration of 1×10^{18} cm⁻³ or more, which has an electronic conductance equivalent to an electric conductivity of 10^{-5} Scm⁻¹ or more at a room temperature.

2. (Previously Presented): A $12\text{SrO} \cdot 7\text{Al}_2\text{O}_3$ compound, which incorporates a negative hydrogen ion (H⁻, H²⁻, H₂⁻) at a concentration of 1×10^{18} cm⁻³ or more, which has an electronic conductance equivalent to an electric conductivity of 10^{-5} Scm⁻¹ or more at a room temperature.

3. (Previously Presented): A mixed crystal compound of $12\text{CaO} \cdot 7\text{Al}_2\text{O}_3$ and $12\text{SrO} \cdot 7\text{Al}_2\text{O}_3$, which incorporates a negative hydrogen ion (H⁻, H²⁻, H₂⁻) at a concentration of 1×10^{18} cm⁻³ or more, which has an electronic conductance equivalent to an electric conductivity of 10^{-5} Scm⁻¹ or more at a room temperature.

4-6. (Cancelled).

7. (Currently Amended): A method of producing the compound as defined in either one of claims 1 to 3 claim 1, comprising subjecting either one selected from the group consisting of a 12CaO · 7Al₂O₃ compound, a 12SrO · 7Al₂O₃ compound, and a mixed crystal compound of

- 2 -

Amendment under 37 CFR §1.111 Attorney Docket No.: 042834

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Art Unit: 1793

 $12\text{CaO} \cdot 7\text{Al}_2\text{O}_3$ and $12\text{SrO} \cdot 7\text{Al}_2\text{O}_3$ to a heat treatment at a temperature of 800°C or more in an atmosphere containing 1000 ppm or more of hydrogen, to thereby clathrate a negative hydrogen ion (H^-, H^{2-}, H_2^-) into said selected compound at a concentration of 1×10^{18} cm⁻³ or more, and further irradiate said selected compound with ultraviolet ray or X-ray.

8-11. (Cancelled).

12. (New): A method of producing the compound as defined in claim 2, comprising subjecting either one selected from the group consisting of a $12\text{CaO} \cdot 7\text{Al}_2\text{O}_3$ compound, a $12\text{SrO} \cdot 7\text{Al}_2\text{O}_3$ compound, and a mixed crystal compound of $12\text{CaO} \cdot 7\text{Al}_2\text{O}_3$ and $12\text{SrO} \cdot 7\text{Al}_2\text{O}_3$ to a heat treatment at a temperature of 800°C or more in an atmosphere containing 1000 ppm or more of hydrogen, to thereby clathrate a negative hydrogen ion (H^-, H^{2-}, H_2^-) into said selected compound at a concentration of 1×10^{18} cm⁻³ or more, and further irradiate said selected compound with ultraviolet ray or X-ray.

13. (New): A method of producing the compound as defined in claim 3, comprising subjecting either one selected from the group consisting of a 12CaO · 7Al₂O₃ compound, a 12SrO · 7Al₂O₃ compound, and a mixed crystal compound of 12CaO · 7Al₂O₃ and 12SrO · 7Al₂O₃ to a heat treatment at a temperature of 800°C or more in an atmosphere containing 1000 ppm or more of hydrogen, to thereby clathrate a negative hydrogen ion (H⁻, H²⁻, H₂⁻) into said

Application No.: 10/511,715 Amendment under 37 CFR §1.111 Art Unit: 1793 Attorney Docket No.: 042834

selected compound at a concentration of 1×10^{18} cm⁻³ or more, and further irradiate said selected compound with ultraviolet ray or X-ray.

14. (New): The compound as defined in any one of claims 1 to 3, wherein the compound is included in a transparent electrode or wiring.

15. (New): The compound as defined in any one of claims 1 to 3, wherein the compound is included in an optically writable and erasable 3-dimensional electronic circuit and 3-dimensional storage element.

16. (New): The compound as defined in any one of claims 1 to 3, wherein the compound is included in a negative-hydrogen-ion-conducting solid-electrolyte.

- 4 -